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Reviews

It in a of the best book. Indeed, it really is play, nevertheless an amazing and interesting literature. It is extremely difficult to leave it before concluding, once you begin to read the book. (Sofia Yundt)

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paperback. Book Condition: New. Ship out in 2 business day, And Fast shipping, Free Tracking number will be provided after the shipment.Pages Number: 288 Publisher: Science Press Pub. Date: 2009-08. This book is a class of professional engineering institutions of higher learning general chemistry curriculum materials. The book is 12 chapters. based on some of the first nine chapters. focuses on chemical thermodynamics. chemical kinetics. chemical equilibrium and the knowledge of the structure of matter; after 3 colloidal. elements and compounds. elemental knowledge application section. Book selection and strive to combine the breadth and depth. as far as possible reflect the views of modern science. Book is strictly the implementation of our legal units of measurement requirements. Reference book with answers and appendices. Contents: Preface Chapter 0 Chapter 1 Introduction 1.1 System of chemical reaction rates. environmental and chemical reaction rates with 1.2 and 1.3 notation impact chemical reaction rate theory of chemical reaction rate 1.4 1.5 a simple series of factors, the reaction rate equations and half-thinking Exercises Chapter 2 questions the energy and direction of chemical reactions 2.1 -2.2 law of energy conservation and transformation of the thermal effect of chemical reactions and calculation of the direction of a chemical reaction with 2.3 Gibbs free energy thinking questions Exercises Chapter 3 Chemical Equilibrium Chemical Equilibrium constant of 3.1 thinking of moving chemical balance 3.2 Exercises Chapter 4 of Title 4.1 a strong electrolyte solution and the effective concentration of electrolyte ionization theory 4.3 4.2 pH of water proton ionization and electrolyte solution ph 4.4 with the ionization balance 4.5 4.6 ion effect and proton acid buffer solution. alkaline reaction with water has a pH of 4.7 with the ionic dissociationFour Satisfaction guaranteed,or money back.



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